Thermo-Physical Properties of Intermediate Temperature Heat Pipe Fluids

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Abstract. Heat pipes are among the most promising technologies for space radiator systems. The paper reports further evaluation of potential heat pipe fluids in the intermediate temperature range of 400 to 700 K in continuation of two recent reports. More thermo-physical property data are examined. Organic, inorganic and elemental substances are considered. The evaluation of surface tension and other fluid properties are examined. Halides are evaluated as potential heat pipe fluids. Reliable data are not available for all fluids and further database development is necessary. Many of the fluids considered are promising candidates as heat pipe fluids. Water is promising as a heat pipe fluid up to 500-550 K. Life test data for thermo-chemical compatibility are almost non-existent.

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Space Power Heat Rejection Technologies

- Needs higher temperature heat rejection systems: 400 700 K considered as an envelope
- Necessity to save on mass and size
- Heat pipes potentially are important components of HRS

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Heat Pipe Technologies - Current Status

- Below 400 K
- Ex. : Commercial and space electronics applications
 - mature technology
- Above 700 K
- High temperature alkali metal heat pipes technologies exist
- Temperature range 400 700 K: Defined as Intermediate Temperature Heat Pipe Technologies
 - Far less developed





Intermediate Temperature Heat Pipes

Current Technology Development Approaches

- Augment the upper operational temperature limit of the ambient temperature heat pipe technology-Ex: Water heat pipes
- Develop new technologies for the intermediate temperature heat pipes

Heat pipe fluids and compatible metals Wick structures

Fabrication technologies

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New in the Current Work

Recent Reports

- Anderson et al. (2004)
- Devarakonda and Olminsky (2004)

Present Work

- Better data
- Better property estimates
- More fluids



Intermediate Temperature Heat Pipe Fluids

Important characteristics to look for

- Maximum operational temperature should be at least 100 K below the critical temperature
- Must be in fluid phase, i.e., melting point below the operational temperature

characteristics. Some elements (Ex. Sulfur and A number of halides seems to have the requisite lodine) are also potential heat pipe fluids.

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Properties - Overall View

- Reliable data are not available for many materials
- Estimates are made if a given property is known for one material in a family.
- If a method is known to work to estimate properties of a family of fluids, that method is used for other families too, based on the known data.

In some cases no data are known. Ex.: The following halides could not be evaluated

- BCl₃, BBr₃, Bl₃
- SiCl₄, SiBr₄, Sil₄

Possible heat pipe fluids: Halides of AI, BI, Ga, Sb, Sn, and Ti

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Surface Tension Estimation

Reid, Prausnitz and Poling (1987) recommendation: properties and normal boiling point are known Based on corresponding states equation- critical First step: Calculate Q

$$Q = 0.1196 \left(1 + \frac{T_{Boil}}{T_{Critical}} \ln(P_{Critical}) \right)$$

$$1 - \frac{T_{Boil}}{T_{Critical}}$$

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Surface Tension Estimation

Second step: Calculate surface tension, σ

$$\sigma = Q \bigg(\frac{P_{Critical}}{Bar} \bigg)^{\frac{2}{3}} \bigg(\frac{T_{Critical}}{K} \bigg)^{\frac{1}{3}} \bigg(1 - \frac{K}{T_{Critical}} \bigg)^{\frac{11}{9}} \frac{N}{m}$$

Better estimation is given by (authors):

$$\sigma_{\text{Halide}} = \frac{2}{3}\sigma$$

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Vapor Pressure

Vapor pressure should not be too high or too low

- SnCl₄ Experimental

AICI₃ Estimated

--- TiCl, Experimental - - SnCl, Estimated

TICI₄

0.03

0.04

-- TiCl, Estimated

SnCl

0.02

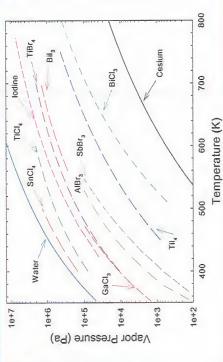
Surface Tension (N/m)

0.01

AC₃

AICI₃ Experimental

Surface Tension



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MASA

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650

009

550

200

450

400

350

300

0.00 L 250

Temperature (K)

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Vapor Pressure

- Water heat pipe technologies are widely available. Hence, water is benchmarked as the high end
- Cesium vapor pressure is too low in this temperature range. Hence, it is considered as the low end.

The vapor pressures of all the fluids evaluated fall within this range.

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Figure of Merit

Definition of Figure of Merit (In "Liquid Transport Factor") Heat Pipe literature, also called, "Merit Number",

whose values are desired to be higher and those desired higher the value of the Merit properties in the numerator Number, the better the fluid is thought to be for a heat denominator, Hence, the Merit Number, M, has those to be lower in the

 $M = \rho_L \cdot \sigma \cdot \lambda$

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Some Promising Fluids

- Antimony Tribromide (Sb Br₃)
- Bismuth Trichloride (Bi Cl₃)

High viscosity but compensated by high surface tension

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700

9

400

300

8

16+9

Temperature (K) 200

Sulfur/lodine

SB

Figure of Merit (W/m²)

Water

Figure of Merit

16+12

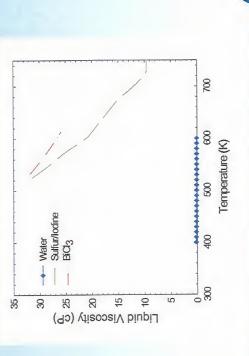
lodine

Ba

SnO4 GaBr3



Viscosity



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Materials Compatibility Testing at NASA Glenn

Tests in progress

Al Br₃ with

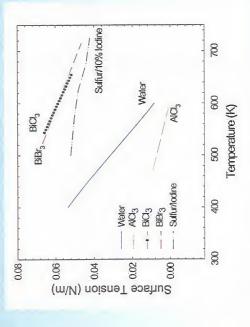
Al 5052, Al 6061, Ti gr2

Tests planned for the near future

Sb Br3 with

Al and Ti

Surface Tension



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Summary

- Halides and some elemental materials evaluated for potential Intermediate Temperature Heat Pipes, 400 to 700 K.
- Properties estimated from known data wherever possible
- Not all possible fluids could be evaluated due to paucity of data
- Some materials identified as potential heat pipe fluids and for thermo-chemical compatibility testing with metals



